Note

A new enzymic method for determination of the number-average degree of polymerization (d.p.n) of linear, amylose-type polysaccharides

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The action pattern of the enzyme phosphorylase shows that maltotetraose is the limit-dextrin for both the degradative and synthetic actions. Phosphorylase requires maltotetraose as the lowest primer for the efficient synthesis of larger molecules and it cannot degrade maltotetraose or smaller saccharides^{1,2}. Moreover, four p-glucose residues remain in the outer chains after the action of phosphorylase on amylopectin³. This action pattern may be used as a basis for the determination of the d.p., of linear, amylose-type polysaccharides. Banks and Greenwood⁴ used a method based on the degradation of amylose chains with beta-amylase. High concentrations of this enzyme produce maltose and p-glucose from chains having even and odd numbers of p-glucose residues, respectively. The d.p., is then calculated from the concentration of p-glucose and maltose produced, on the assumption that the polysaccharide contains an equal number of molecules having even and odd numbers of p-glucose residues.

The use of phosphorylase instead of beta-amylase eliminates this assumption because phosphorylase will degrade every chain to maltotetraose (G_4) in addition to D-glucose 1-phosphate (G-1-P). If the amount of G_4 produced can be determined and related to total polysaccharide concentration, then $\overline{d.p.}_n$ is given by $[G_t]/[G_4]$, where $[G_t]$ and $[G_4]$ are the molar concentrations of total D-glucose and maltotetraose, respectively.

TABLE I $\overline{D.P.}_n$ of the linear glucans

	G_6	G ₇	Amylose samples				
			1	2	3	4	5
D.p., by present method	6.1	7.1	165	215	315	586	830
D.p.n by Banks and Greenwood method ⁴		a	161	218	312	5 72	818

[&]quot;Not determined.

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Table I shows the values of $\overline{d.p.}_n$ determined by the present and previous methods. The present method has two advantages over the method of Banks and Greenwood. First, every chain produces four molecules of p-glucose; therefore, this method is 8 times more sensitive. Second, very small amounts (0.5 mg) of poly-saccharide are required for the determination. Moreover, the method may be applied to determine the average chain-length $(\overline{c.l.}_n)$ of the branched, amylopectin-type polysaccharides, provided they are completely debranched before degradation with phosphorylase.

EXPERIMENTAL

Samples of linear amylose were obtained by leaching potato-starch granules⁵ followed by degradation with bacterial alpha-amylase. Samples were taken out at intervals and the amylose was precipitated as the amylose–1-butanol complex. The samples were redissolved in water.

Maltohexaose (G_6) and maltoheptaose (G_7) were prepared by incubating amylose with bacterial alpha-amylase for a longer period, and G_6 and G_7 were separated chromatographically. The samples were purified by rechromatography and freeze-dried.

Potato phosphorylase (EC 2.4.1.1), prepared as described by Lee⁶, was free from alpha-amylase activity. Amyloglucosidase [$(1\rightarrow4)$ - α -D-glucan glucohydrolase, EC 3.2.1.3], D-glucose oxidase (EC 1.1.3.4), peroxidase (EC 1.11.1.7), bacterial alpha-amylase (EC 3.2.1.1), and beta-amylase (EC 3.2.1.2) were obtained from Calbiochem.

Preliminary experiments. — G_4 can be determined by treatment with amyloglucosidase and assay of the D-glucose released (D-glucose oxidase method⁴). The total concentration (G_t) of the polysaccharide can also be obtained enzymically by the same method after acid hydrolysis of the ester phosphate group. The accuracy of the method depends upon the determination of D-glucose in the presence of large amounts of D-glucose 1-phosphate (G-1-P). There are two possible sources of error: (1) G-1-P might be degraded by the amyloglucosidase, (2) the determination of D-glucose involves two impure enzymes, D-glucose oxidase and peroxidase, and G-1-P might be hydrolysed by an impurity. The following experiment was designed to check this possibility.

A solution (5 ml) of G-1-P (1 mg/ml) containing 0.1M acetate buffer (pH 4.8, 1 ml) and amyloglucosidase solution (0.1 ml, 1-2 units) was incubated at 37° for 22 h. Assay by the p-glucose oxidase method then revealed that no p-glucose had been formed. Thus, amyloglucosidase can be used for the selective degradation of G_4 in the presence of G-1-P. Moreover, the p-glucose oxidase reagent did not interfere in the assay. Also, the assay of p-glucose was not affected by the presence of G-1-P.

Determination of $\overline{d.p.}_n$. — (a) Determination of G_4 . Digests for G_6 contained a solution (0.5 ml) of G_6 (1 mg/ml), 0.1M phosphate buffer (pH 6.3, 0.2 ml), and phosphorylase solution (0.2 ml, 350 units/ml)⁶. Similar digests were set up for G_7

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and amylose. The digests were incubated at 37° for 18 h and the pH was then adjusted to 4.8 by the addition of 0.1m acetate buffer (pH 4.8, 0.5 ml). Amyloglucosidase solution (0.1 ml, 1.2 units)⁴ was added, and each digest was diluted to 5 ml with distilled water and incubated at 37° for 4 h. The p-glucose produced was determined by the p-glucose oxidase method.

(b) Determination of G_t . A portion (1 ml) of each of the above digests was hydrolysed with $0.25 \text{M H}_2 \text{SO}_4$ (1 ml) at $\sim 100^\circ$ for 1.5 h, the acid was then neutralized, and the D-glucose produced was determined as in (a).

 $\overline{D.p._n}$ was also determined by degrading the samples with beta-amylase, as described by Banks and Greenwood⁴.

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REFERENCES

- 1 S. HESTRIN, J. Biol. Chem., 179 (1949) 943-955.
- 2 W. J. WHELAN AND J. M. BAILEY, Biochem. J., 58 (1954) 560-569.
- 3 G. J. WALKER AND W. J. WHELAN, Biochem. J., 76 (1961) 257-263.
- 4 W. BANKS AND C. T. GREENWOOD, Carbohyd. Res., 6 (1968) 177-183.
- 5 W. Banks, C. T. Greenwood, and J. Thompson, Makromol. Chem., 31 (1959) 197-213.
- 6 YA-PIN LEE, Biochim. Biophys. Acta, 43 (1960) 18-24.
- 7 W. BANKS, C. T. GREENWOOD, AND K. M. KHAN, Carbohyd. Res., 12 (1970) 79-87.